Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

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## Key indicators

Single-crystal X-ray study
$T=193 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.002 \AA$
$R$ factor $=0.054$
$w R$ factor $=0.123$
Data-to-parameter ratio $=17.4$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## rac-(tert-Butylamino)(diisopropylamino)-(9-fluorenyl)borane

In the crystal structure, the title compound, $\mathrm{C}_{23} \mathrm{H}_{33} \mathrm{BN}_{2}$, exhibits approximate $C_{\mathrm{s}}$ symmetry, with $\mathrm{B}-\mathrm{N}$ bond lengths of 1.414 (2) and 1.419 (2) $\AA$, consistent with $\mathrm{B}=\mathrm{N}$ double bonds. This is also reflected by the trigonal planes around the B and butylamino N atoms, which are virtually coplanar.

## Comment

Ligand precursors related to the title compound rac-(tert-butylamino)(diisopropylamino)(9-fluorenyl)borane, (I), have been previously employed for the synthesis of boron-bridged constrained geometry complexes. These complexes exhibit activity for the polymerization of ethene and higher $\alpha$-olefins when activated with MAO (methylaluminoxane) (Braunschweig et al., 2000, 2004). In comparative studies, a strong influence of the ligand substitution patterns in corresponding complexes has been demonstrated on both catalyst activity and polymerization characteristics (Stevens 1994), justifying the synthesis of a series of boron-bridged analogues.

(I)

The structure of (I) (Fig. 1) in the solid state largely resembles that of its corresponding $N$-phenyl-substituted counterpart $\left(\eta^{1}-\mathrm{C}_{13} \mathrm{H}_{9}\right) \mathrm{B}\left(\mathrm{N}^{\mathrm{i}} \mathrm{Pr}_{2}\right) \mathrm{N}(\mathrm{H}) \mathrm{Ph}$ (Braunschweig et al., 2003). Compound (I) adopts approximate $C_{\mathrm{s}}$ symmetry in the crystal structure. Again, both the B1 and N2 centres adopt trigonal planar geometries with the central atoms lying only 0.002 ( B 1 ) and $0.001 \AA(\mathrm{~N} 2)$ out of the planes of their respective substituents. These two trigonal planes are virtually coplanar with torsion angles of $5.5(2)(\mathrm{C} 18-\mathrm{N} 2-\mathrm{B} 1-\mathrm{N} 1)$ and $5.3(2)^{\circ}(\mathrm{C} 21-\mathrm{N} 2-\mathrm{B} 1-\mathrm{C} 9)$. The $\mathrm{B} 1-\mathrm{N} 1$ and $\mathrm{B} 2-\mathrm{N} 2$ bond lengths of 1.414 (2) and 1.419 (2) $\AA$, respectively, are equivalent within experimental error, indicating equal amounts of $\pi$-contribution from the two amino groups. The N2-bound isopropyl group syn to the fluorenyl moiety is oriented such that the methine proton is directed towards the middle of the five-membered ring, with $\mathrm{H}-\pi$ distances to the centres of the $\mathrm{C} 2=\mathrm{C} 10$ and $\mathrm{C} 8=\mathrm{C} 3$ aromatic bonds of 2.561
and $2.558 \AA$, respectively. No short intermolecular contacts are observed.

## Experimental

The title compound was prepared by a modified literature procedure (Braunschweig et al., 2003) by sequential reaction of ${ }^{i} \mathrm{Pr}_{2} \mathrm{NBCl}_{2}$ $(4.55 \mathrm{~g}, 25.0 \mathrm{mmol})$ with lithium fluorenide $(4.30 \mathrm{~g}, 25.0 \mathrm{mmol})$ and lithium tert-butylamide $(1.98 \mathrm{~g}, 25.0 \mathrm{mmol})$ in a hexane-toluene mixture. Two subsequent recrystallization steps from hexane at ambient temperature and from toluene at ambient temperature yielded the pure target compound (I) ( $1.18 \mathrm{~g}, 13.6 \%$ ). Crystals of (I) suitable for single-crystal X-ray diffraction were selected directly from the sample as prepared.

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{23} \mathrm{H}_{33} \mathrm{BN}_{2} \\
& M_{r}=348.32 \\
& \text { Monoclinic, } P 2_{1} / n \\
& a=9.2771(5) \AA \\
& b=16.0667(9) \AA \\
& c=14.5453(8) \AA \\
& \beta=97.0340(10)^{\circ} \\
& V=2151.7(2) \AA^{3} \\
& Z=4
\end{aligned}
$$

$$
\begin{aligned}
& D_{x}=1.075 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \text { Cell parameters from } 9516 \\
& \quad \text { reflections } \\
& \theta=2.5-25.4^{\circ} \\
& \mu=0.06 \mathrm{~mm}^{-1} \\
& T=193(2) \mathrm{K} \\
& \text { Block, colourless } \\
& 0.35 \times 0.25 \times 0.21 \mathrm{~mm}
\end{aligned}
$$

## Data collection

| Bruker SMART CCD area-detector | 4269 independent reflections |
| :--- | :--- |
| $\quad$ diffractometer | 3774 reflections with $I>2 \sigma(I)$ |
| $\varphi$ and $\omega$ scans | $R_{\text {int }}=0.042$ |
| Absorption correction: multi-scan | $\theta_{\max }=26.1^{\circ}$ |
| $\quad(S A D A B S ;$ Sheldrick, 2002) | $h=-11 \rightarrow 11$ |
| $T_{\min }=0.954, T_{\max }=0.987$ | $k=-19 \rightarrow 19$ |
| 35746 measured reflections | $l=-17 \rightarrow 17$ |

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.054$
$w R\left(F^{2}\right)=0.123$
$S=1.12$
4269 reflections
246 parameters
H atoms treated by a mixture of independent and constrained refinement

All H atoms were visible in difference Fourier maps. H atoms bonded to N atoms were refined. H atoms bonded to C atoms were subsequently placed at idealized positions and treated as riding atoms, with $\mathrm{C}-\mathrm{H}=0.98\left(\mathrm{CH}_{3}\right), 0.95$ (aromatic CH$)$ and $1.00 \AA$ (aliphatic CH ). $U_{\text {eq }}$ values for primary H atoms were fixed at 1.5 times and tertiary H atoms at 1.2 times the $U_{\mathrm{eq}}$ of the attached C atom.


Figure 1
A view of (I), with displacement ellipsoids drawn at the $30 \%$ probability level.

Data collection: SMART-NT (Bruker, 2001); cell refinement: SAINT-Plus-NT (Bruker, 2001); data reduction: SAINT-Plus-NT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: XP in SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: $X P$ in $S H E L X T L$.

The authors thank DFG and EPSRC for financial support.

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